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#### Key indicators

Single-crystal X-ray study

$T = 173\text{ K}$

Mean  $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$

$R$  factor = 0.048

$wR$  factor = 0.119

Data-to-parameter ratio = 14.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## *N*-{[1-(5-Benzylsulfanyl)-1,3,4-oxadiazol-2-yl]-ethyl}-4-chlorobenzenesulfonamide

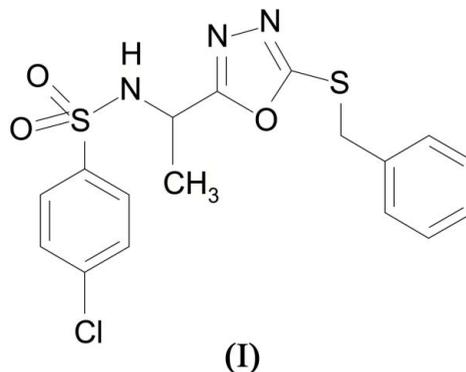
The crystal structure of the title compound,  $\text{C}_{17}\text{H}_{16}\text{ClN}_3\text{O}_3\text{S}_2$ , contains hydrogen-bonded chains lying along the  $b$  axis. In the molecule, the benzene rings lie almost parallel to each other and the heterocyclic oxadiazole ring is oriented nearly perpendicular to the benzene rings.

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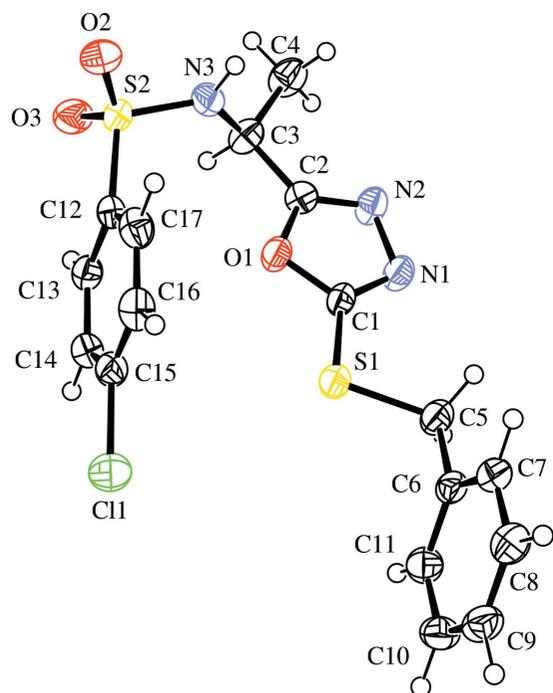
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#### Comment

Benzenesulfonamides and 1,3,4-oxadiazole derivatives have been reported to possess significant biological activities, such as antimicrobial, anti-HIV, insulin-releasing antidiabetic, carbonic anhydrase inhibitory, high-ceiling diuretic, anti-thyroid, antitumour, *etc.* (Nishimori *et al.*, 2006; Turner, 2002; Supuran & Scozzafava, 2000, 2001, 2003; Masereel *et al.*, 2002; Singh *et al.*, 1997; Khanum *et al.*, 2005). In continuation of our interest in the chemical and pharmacological properties of benzenesulfonamides and 1,3,4-oxadiazole derivatives (Zareef *et al.*, 2006), we have synthesized a series of new compounds. We report the structure of the title compound, (I), in this paper.



Molecules of (I) (Fig. 1) form hydrogen-bonded chains along the  $b$  axis *via* strong  $\text{N}-\text{H}\cdots\text{N}$  hydrogen bonds (Fig. 2); details of the hydrogen-bonding geometry are provided in Table 2. Unlike the structure of 4-methyl-*N*-[1-(5-mercapto-1,3,4-oxadiazol-2-yl)propyl]benzenesulfonamide (Zareef *et al.*, 2006), which is closely related to (I), the O atoms bonded to atom S2 are not involved in hydrogen bonds. The oxadiazole ring is essentially planar and the benzene rings lie approximately perpendicular to its mean plane: the angle between the mean planes of the oxadiazole and chlorophenyl rings is  $71.61(10)^\circ$ , while that between the oxadiazole and benzyl ring is  $84.10(9)^\circ$ . The benzene rings lie almost parallel to each other, the angle between the mean planes being  $6.25(18)^\circ$ . The molecular dimensions in (I) are as expected. A search of the Cambridge Structural Database (2006 release; Allen,



**Figure 1**  
The molecular structure of (I), with displacement ellipsoids plotted at the 50% probability level.

2002) for the 5-thio-1,3,4-oxadiazole skeleton yielded only five entries [refcodes AVULIM (Ozturk *et al.*, 2004), AVULUY (Du *et al.*, 2004), YITMUJ (Ziyaev *et al.*, 1992), IZAJEY (Qiu & Xu, 2004) and UGOBEX (Zhang *et al.*, 2002)]; the first three compounds are thiones with S=C double bonds, while IZAJEY is a thioglucofuranoside derivative with S—C distances similar to those found in (I). Three-dimensional coordinates for UGOBEX, a benzylsulfanyloxadiazole derivative of a triazole, are not available in the CSD.

## Experimental

Compound (I) was synthesized in four steps. D-2-(4-Chlorophenylsulfonamido)propanoic acid was esterified with ethanol in an acidic medium using the standard method (Furniss *et al.*, 1978). A mixture of the resulting ethyl-2-(4-chlorophenylsulfonamido)propanoate (10 mmol) and hydrazine monohydrate (80%) in absolute ethanol (60 ml) was refluxed for 9 h. The excess solvent was distilled off and the residue was filtered off, washed with water and recrystallized from 60% aqueous ethanol to yield 2-(4-chlorophenylsulfonamido)propane hydrazide. 4-Chloro-*N*-[1-(5-mercapto-1,3,4-oxadiazol-2-yl)ethyl]benzenesulfonamide was obtained by the reaction of this hydrazide (5.5 mmol) in absolute ethanol (80 ml) with carbon disulfide (6.6 mmol) and aqueous potassium hydroxide (5.5 mmol) at reflux temperature for 17 h. Finally, 4-chloro-*N*-[1-(5-mercapto-1,3,4-oxadiazol-2-yl)ethyl]benzenesulfonamide (0.75 mmol), Et<sub>3</sub>N (0.22 mmol) and a catalytic amount of 4-(dimethylamino)pyridine (DMAP) (25 mg) were stirred in dry CH<sub>2</sub>Cl<sub>2</sub> (25 ml) for 15 min. Benzyl bromide (0.8 mmol) was added and the mixture was stirred for 5 h at 323–343 K. The reaction mixture was washed with dilute HCl, brine and water, and dried over Na<sub>2</sub>SO<sub>4</sub>

(anhydrous). The excess solvent was distilled off and the product was recrystallized from aqueous ethanol to obtain the pure title compound, (I). Crystals suitable for crystallographic study were grown from a solution of (I) in ethanol by slow evaporation at room temperature.

### Crystal data

C<sub>17</sub>H<sub>16</sub>ClN<sub>3</sub>O<sub>3</sub>S<sub>2</sub>  
*M<sub>r</sub>* = 409.90  
 Monoclinic, *P*2<sub>1</sub>/*c*  
*a* = 10.353 (3) Å  
*b* = 7.592 (3) Å  
*c* = 23.655 (10) Å  
 β = 100.329 (19)°  
*V* = 1829.2 (12) Å<sup>3</sup>

*Z* = 4  
*D<sub>x</sub>* = 1.488 Mg m<sup>-3</sup>  
 Mo Kα radiation  
 μ = 0.46 mm<sup>-1</sup>  
*T* = 173 (2) K  
 Needle, colourless  
 0.25 × 0.06 × 0.03 mm

### Data collection

Nonius KappaCCD area-detector diffractometer  
 ω and φ scans  
 Absorption correction: multi-scan (SORTAV; Blessing, 1997)  
*T<sub>min</sub>* = 0.894, *T<sub>max</sub>* = 0.986

11365 measured reflections  
 3314 independent reflections  
 2422 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.063  
 θ<sub>max</sub> = 25.3°

### Refinement

Refinement on *F*<sup>2</sup>  
*R*[*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.048  
*wR*(*F*<sup>2</sup>) = 0.119  
*S* = 1.05  
 3314 reflections  
 236 parameters  
 H-atom parameters constrained

*w* = 1/[σ<sup>2</sup>(*F<sub>o</sub>*<sup>2</sup>) + (0.045*P*)<sup>2</sup> + 1.95*P*]  
 where *P* = (*F<sub>o</sub>*<sup>2</sup> + 2*F<sub>c</sub>*<sup>2</sup>)/3  
 (Δ/σ)<sub>max</sub> = 0.001  
 Δρ<sub>max</sub> = 0.71 e Å<sup>-3</sup>  
 Δρ<sub>min</sub> = -0.35 e Å<sup>-3</sup>

**Table 1**

Selected geometric parameters (Å, °).

C11—C15	1.736 (3)	O1—C1	1.360 (3)
S1—C1	1.723 (3)	O1—C2	1.365 (4)
S1—C5	1.821 (3)	N1—C1	1.296 (4)
S2—O3	1.425 (2)	N1—N2	1.418 (4)
S2—O2	1.437 (3)	N2—C2	1.283 (4)
S2—N3	1.614 (3)	N3—C3	1.449 (4)
S2—C12	1.760 (3)		
C1—S1—C5	98.76 (15)	N3—S2—C12	108.00 (14)
O3—S2—O2	120.08 (15)	C1—O1—C2	103.0 (2)
O3—S2—N3	109.84 (16)	C1—N1—N2	106.2 (2)
O2—S2—N3	103.25 (15)	C2—N2—N1	106.0 (3)
O3—S2—C12	106.85 (14)	C3—N3—S2	124.2 (2)
O2—S2—C12	108.33 (16)		

**Table 2**

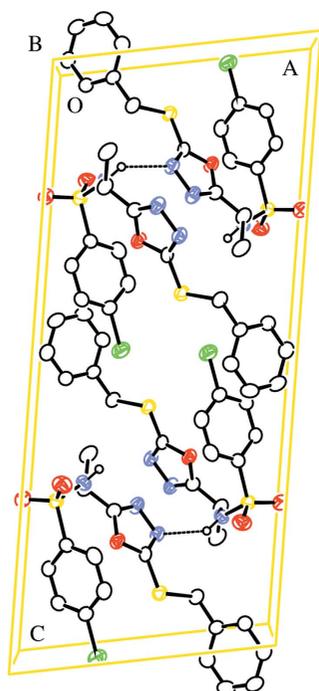
Hydrogen-bond geometry (Å, °).

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
N3—H3A···N1 <sup>i</sup>	0.88	2.19	2.936 (4)	143

Symmetry code: (i)  $-x + 1, y + \frac{1}{2}, -z + \frac{3}{2}$ .

H atoms were included in the refinement in geometrically idealized positions, with N—H = 0.88 and C—H = 0.95–1.00 Å, and *U*<sub>iso</sub> = 1.2*U*<sub>eq</sub>(C,N).

Data collection: COLLECT (Nonius, 1998); cell refinement: HKL DENZO (Otwinowski & Minor, 1997); data reduction: SCALEPACK (Otwinowski & Minor, 1997); program(s) used to solve structure: SAPIRI (Fan, 1991); program(s) used to refine structure:



**Figure 2**

The packing of (I), showing the N—H···N hydrogen bonds (dashed lines). H atoms not involved in hydrogen bonding have been omitted.

*SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP II* (Johnson, 1976); software used to prepare material for publication: *SHELXL97* (Sheldrick, 1997).

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